R20

Q.P. Code: 20HS0804

Reg. No:

SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR

(AUTONOMOUS)

B.Tech I Year II Semester Regular Examinations November-2021 ENGINEERING CHEMISTRY

(Common to CE & AGE)

	(Common to CE & MGE)		
	Time: 3 hours	Max.	Marks: 60
	(Answer all Five Units $5 \times 12 = 60$ Marks)		
	UNIT-I		
1	a What is priming and foaming?	L1	6M
	b Explain the process of scale and sludge formation in boilers.	L2	6M
	OR		
2	Describe the Ion exchange process for demineralization of water? What are the	L3	12M
	advantages and disadvantages of ion exchange process?		
	UNIT-II		
3	a What is primary Battery? Write a note on Zinc-air battery.	L1	6M
	b Describe the Construction and Working of Methanol – Oxygen Fuel cell.	L3	6M
	OR		
4	a Write a note on sacrificial anodic protection?	L1	4M
	b What is electroplating? Explain electroplating of Nickel and copper?	L2	8M
	UNIT-III		
5	Explain the analysis of Coal (Proximate and Ultimate) With its Significance.	L2	12M
	OR		
6	a What is functionality of monomer?	L1	4M
	b Write the preparation, properties and application of Buna-S rubber,	L2	8M
	Buna-N rubber and Thiokol rubber.		
	UNIT-IV		
7	Define Viscosity? Determine the viscosity of lubricating oil by Redwood Viscometer.	L2	12M
	OR		
8	a What is cement? How do you classify the cement?	L1	6 M
	b Explain in detail about setting and hardening of Portland cement?	L2	6 M
	UNIT-V		
9	Write a note on any one chemical and electrochemical methods of preparation of	L1	12M
	nano metals.		
OR			
10	a Explain the BET Equation.	L2	6M
	b What are the factors influencing Adsorption of gases on solids.	L1	6M